Table 2. Bond lengths ( $\AA$ ) and angles $\left({ }^{\circ}\right)$

| $\mathrm{C}(2)-\mathrm{N}(1) \quad 1.377$ | 1.377 (2) | $\mathrm{C}(5)-\mathrm{N}(1) \quad 1.3$ | 1.379 (2) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(3)-\mathrm{C}(2) \quad 1.5$ | . 524 (2) | $\mathrm{O}(2)-\mathrm{C}(2) \quad 1.2$ | 1.210 (2) |
| $\mathrm{C}(4)-\mathrm{C}(3) \quad 1.53$ | 1.536 (2) | $\mathrm{C}(31)-\mathrm{C}(3) \quad 1.5$ | 1.533 (2) |
| $\mathrm{C}(5)-\mathrm{C}(4) \quad 1.510$ | 1.510 (2) | $\mathrm{C}(41)-\mathrm{C}(4) \quad 1.5$ | 1.525 (3) |
| $\mathrm{O}(5)-\mathrm{C}(5) \quad 1.21$ | 1.217 (2) | $\mathrm{C}(32)-\mathrm{C}(31) \quad 1.5$ | 1.514 (3) |
| $\mathrm{O}(33)-\mathrm{C}(31) \quad 1.4$ | 1.434 (2) | $\mathrm{C}(34)-\mathrm{O}(33) \quad 1.4$ | 1.426 (2) |
| $\mathrm{C}(5)-\mathrm{N}(1)-\mathrm{C}(2)$ | 113.8 (1) | $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{N}(1)$ | 107.9 (1) |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{N}(1)$ | 123.8 (2) | $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{C}(3)$ | 128.3 (2) |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(2)$ | 104.6 (1) | $\mathrm{C}(31)-\mathrm{C}(3)-\mathrm{C}(2)$ | 110.4 (1) |
| $\mathrm{C}(31)-\mathrm{C}(3)-\mathrm{C}(4)$ | 113.7 (1) | $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(3)$ | 104.5 (1) |
| $\mathrm{C}(41)-\mathrm{C}(4)-\mathrm{C}(3)$ | 114.3 (2) | $\mathrm{C}(41)-\mathrm{C}(4)-\mathrm{C}(5)$ | 109.7 (1) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(1)$ | 108.4 (1) | $\mathrm{O}(5)-\mathrm{C}(5)-\mathrm{N}(1)$ | 124.3 (1) |
| $\mathrm{O}(5)-\mathrm{C}(5)-\mathrm{C}(4)$ | 127.3 (1) | $\mathrm{C}(32)-\mathrm{C}(31)-\mathrm{C}(3)$ | 114.0 (2) |
| $\mathrm{O}(33)-\mathrm{C}(31)-\mathrm{C}(3)$ | (3) 104.5 (1) | $\mathrm{O}(33)-\mathrm{C}(31)-\mathrm{C}(32)$ | 112.9 (2) |
| $\mathrm{C}(34)-\mathrm{O}(33)-\mathrm{C}(31$ | 1) $115 \cdot 0$ (2) |  |  |

parameters were refined freely with individual isotropic temperature factors. The terminal value of $R_{w}$ was 0.050 with $R=0.050$.* Weights were given by $w=$ $k\left[\sigma^{2}\left(F_{o}\right)+g\left(F_{o}^{2}\right)\right]^{-1}$, where $g$ refined to 0.000380 . Complex neutral-atom scattering factors were employed (Cromer \& Waber, 1965; Cromer \& Liberman, 1970). Table 1 lists the final non-hydrogen atom coordinates, Table 2 the bond lengths and angles. Calculations were carried out with SHELX (Sheldrick, 1976) and local programs. Fig. 1 was drawn with RSPLOT (W. S. Sheldrick).

[^0]Discussion. Molecules of (1b) build centrosymmetric dimers through $\mathrm{N}(1)-\mathrm{H} \cdots \mathrm{O}(5)$ hydrogen bonds of length $2.88(1) \AA . \mathrm{N}(1), \mathrm{C}(2), \mathrm{C}(3)$ and $\mathrm{C}(4)$ are coplanar: distances from the least-squares plane are $\mathrm{N}(1) 0.002, \mathrm{C}(2)-0.002, \mathrm{C}(3) 0.001, \mathrm{C}(5)-0.001 \AA$. $\mathrm{C}(4)$ is displaced $-0.134, \mathrm{O}(2)-0.008$ and $\mathrm{O}(5)$ $-0.055 \AA$ from this plane. Bond angles at $C(2)$ and $C(3)$ are similar to those at $C(5)$ and $C(4)$ respectively. The torsion angles $\mathrm{H}(31)-\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(41)$ and $\mathrm{H}(41)-\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(31)$ are respectively $3(1)$ and $3(1)^{\circ}$. A torsion angle of $127.6(2)^{\circ}$ is observed for C(31)-C(63)-C(4)-C(41). It may, therefore, be concluded that the five-membered ring is relatively non-strained for $\mathrm{C}(31)$ and $\mathrm{C}(41)$ in a trans position to one another. A similar observation was made for the model compound ( $R R S-S S R$ )-2-[1-(ethylsulphonyl)-ethyll-3-methylsuccinimide (Lotter, Klein, Rüdiger \& Scheer, 1977), in which C(31) and C(41) are likewise trans to one another. A comparison of the geometry of (1b) with that of (1c), for which $\mathrm{C}(31)$ and $\mathrm{C}(41)$ are cis to one another, is made in the following paper (Sheldrick, 1981).

## References

Cromer, D. T. \& Llberman, D. (1970). J. Chem. Phys. 53, 1891-1898.
Cromer, D. T. \& Waber, J. T. (1965). Acta Cryst. 18, 104-109.
Gossauer, A. \& Hinze, J. -P. (1981). Hoppe-Seyler's Z. Physiol. Chem. In the press.
Lotter, H., Klein, G., Rüdiger, W. \& Scheer, H. (1977). Tetrahedron Lett. pp. 2317-2320.
Sheldrick, G. M. (1976). SHELX. A program for crystal structure determination. Univ. of Cambridge, England. Sheldrick, W. S. (1981). Acta Cryst. B37, 300-302.

# (SRS-RSR)-2-(1-Methoxyethyl)-3-methylsuccinimide 

By W. S. Sheldrick<br>Gesellschaft für Biotechnologische Forschung mbH, Mascheroder Weg 1, D-3300 Braunschweig-Stöckheim, Federal Republic of Germany

(Received 25 July 1980; accepted 6 October 1980)

Abstract. $\mathrm{C}_{8} \mathrm{H}_{13} \mathrm{NO}_{3}$, monoclinic, $\quad P 2_{1} / c, \quad a=$ 12.984 (4),$b=7.559$ (3), $c=9.734$ (3) $\AA, \beta=$ 97.77 (4) ${ }^{\circ}, M_{r}=171.2, Z=4, D_{x}=1.20 \mathrm{Mg} \mathrm{m}^{-3}$; $R=0.057$ and $R_{w}=0.051$ for 1420 observed reflexions. The relative configuration of the three chiral centres is $S R S$ or $R S R$. The molecules build centrosymmetric

0567-7408/81/010300-03\$01.00
dimers through $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds of length 2.91 (1) $\AA$. The five-membered ring is markedly twisted with $C(3)$ and $C(4)$ displaced respectively -0.114 and $0 \cdot 180 \AA$ from the plane of the remaining ring atoms. A torsion angle of $23.3(2)^{\circ}$ is observed between the cis-substituted C(3) and C(4).
(c) 1981 International Union of Crystallography

Introduction. In order to elucidate the configurations of methanol adducts of the bile pigment phycocyanobilin dimethyl ester, all four racemic stereoisomers (1a)-(1d) of 2-(1-methoxyethyl)-3-methylsuccinimide were synthesized by Gossauer \& Hinze (1981). The previous paper (Sheldrick, 1981) reported the structure of (1b), an isomer with trans-substituted $\mathrm{C}(3)$ and $\mathrm{C}(4)$. This paper reports the structure of the cis isomer (1c), for which pronounced ring strain would be expected, as a result of steric contacts between the cis substituents.


Cell dimensions were determined by a least-squares fit to settings for 15 reflexions $\pm(h k l)$ on a Syntex $P 2_{1}$ diffractometer $(\mathrm{Cu} K \alpha$ radiation, $\lambda=1.54178 \AA)$. Data collection was carried out in the $\theta-2 \theta$ mode ( $2 \theta \leq$ $135^{\circ}$ ) with graphite-monochromated $\mathrm{Cu} K \alpha$ radiation. No absorption correction was applied $[\mu(\mathrm{Cu} K \alpha)=$ $0.68 \mathrm{~mm}^{-1}$ ]. After application of the observation criterion $F \geq 2 \cdot 5 \sigma(F)$, 1420 unique reflexions were retained for use in the analysis. The structure was solved by direct methods and refined by full-matrix least squares with anisotropic temperature factors for the non-hydrogen atoms. The H atom positional parameters were refined freely and, with the exception of the methyl protons of $\mathrm{C}(34)$, were assigned individual isotropic temperature factors. The terminal value of $R_{w}$ was 0.051 with $R=0.057$.* Weights were given by the counting statistics. Complex neutral-atom

[^1]Table 1. Positional parameters and equivalent isotropic thermal parameters

|  | $x$ | $y$ | $z$ | $\begin{gathered} U_{\mathrm{eq}} \\ \left(\AA^{2} \times 10^{3}\right) \dagger \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)$ | 0.3771 (1) | 0.5565 (2) | 0.3921 (2) | 62 |
| C(2) | 0.2792 (2) | $0 \cdot 6089$ (3) | $0 \cdot 3388$ (2) | 64 |
| $\mathrm{O}(2)$ | $0 \cdot 2185$ (1) | $0 \cdot 6807$ (2) | 0.4042 (1) | 91 |
| C(3) | $0 \cdot 2626$ (2) | $0 \cdot 5580$ (3) | $0 \cdot 1846$ (2) | 64 |
| C(31) | 0.1842 (2) | 0.4061 (4) | $0 \cdot 1607$ (3) | 76 |
| C(32) | 0.0736 (3) | $0 \cdot 4556$ (9) | $0 \cdot 1827$ (5) | 127 |
| $\mathrm{O}(33)$ | 0.2246 (1) | 0.2662 (2) | 0.2484 (1) | 81 |
| C(34) | 0.1866 (4) | 0.0933 (5) | 0.2051 (5) | 128 |
| C(4) | 0.3727 (2) | 0.5147 (3) | $0 \cdot 1542$ (2) | 57 |
| C(41) | 0.3885 (2) | 0.3636 (4) | 0.0552 (3) | 76 |
| C(5) | 0.4348 (2) | 0.4877 (3) | 0.2963 (2) | 58 |
| $\mathrm{O}(5)$ | $0 \cdot 5208$ (1) | 0.4247 (2) | 0.3230 (1) | 75 |

Table 2. Bond lengths ( $\AA$ ) and angles $\left({ }^{\circ}\right)$

| $\mathrm{C}(2)-\mathrm{N}(1)$ | $1.365(3)$ | $\mathrm{C}(5)-\mathrm{N}(1)$ | $1.375(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O}(2)-\mathrm{C}(2)$ | $1.207(3)$ | $\mathrm{C}(3)-\mathrm{C}(2)$ | $1.536(3)$ |
| $\mathrm{C}(31)-\mathrm{C}(3)$ | $1.531(4)$ | $\mathrm{C}(4)-\mathrm{C}(3)$ | $1.534(3)$ |
| $\mathrm{C}(32)-\mathrm{C}(31)$ | $1.526(5)$ | $\mathrm{O}(33)-\mathrm{C}(31)$ | $1.415(3)$ |
| $\mathrm{C}(34)-\mathrm{O}(33)$ | $1.440(4)$ | $\mathrm{C}(41)-\mathrm{C}(4)$ | $1.526(3)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)$ | $1.518(3)$ | $\mathrm{O}(4)-\mathrm{C}(5)$ | $1.209(3)$ |
| $\mathrm{C}(5)-\mathrm{N}(1)-\mathrm{C}(2)$ | $114.6(2)$ | $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{N}(1)$ | $124.9(2)$ |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{N}(1)$ | $107.4(2)$ | $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{O}(2)$ | $127.8(3)$ |
| $\mathrm{C}(31)-\mathrm{C}(3)-\mathrm{C}(2)$ | $109.8(2)$ | $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(2)$ | $103.4(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(31)$ | $115.4(2)$ | $\mathrm{C}(32)-\mathrm{C}(31)-\mathrm{C}(3)$ | $114.5(3)$ |
| $\mathrm{O}(33)-\mathrm{C}(31)-\mathrm{C}(3)$ | $106 \cdot 5(2)$ | $\mathrm{O}(33)-\mathrm{C}(31)-\mathrm{C}(32)$ | $112.2(3)$ |
| $\mathrm{C}(34)-\mathrm{O}(33)-\mathrm{C}(31)$ | $114.7(3)$ | $\mathrm{C}(41)-\mathrm{C}(4)-\mathrm{C}(3)$ | $119.3(2)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(3)$ | $104.3(2)$ | $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(41)$ | $112.0(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(1)$ | $107 \cdot 2(2)$ | $\mathrm{O}(5)-\mathrm{C}(5)-\mathrm{N}(1)$ | $125 \cdot 1(2)$ |
| $\mathrm{O}(5)-\mathrm{C}(5)-\mathrm{C}(4)$ | $127.6(2)$ |  |  |



Fig. 1. The dimer of (lc).
scattering factors were employed (Cromer \& Waber, 1965; Cromer \& Liberman, 1970). Table 1 lists the final non-hydrogen atom coordinates, Table 2 the bond lengths and angles. Calculations were carried out with SHELX (Sheldrick, 1976) and local programs. Fig. 1 was drawn by RSPLOT (W. S. Sheldrick).

Discussion. As was observed for ( $1 b$ ), molecules of (1c) build centrosymmetric dimers through $\mathrm{N}(1)-\mathrm{H} \cdots \mathrm{O}(5)$ hydrogen bonds. The $\mathrm{N}(1) \cdots \mathrm{O}(5)$ distance of 2.91 (1) $\AA$ in (1c) is $0.03 \AA$ longer than that in (1b). A marked twisting of the five-membered ring is observed for (1c). Distances from the least-squares plane through $\mathrm{N}(1), \mathrm{C}(2)$ and $\mathrm{C}(5)$ are: $\mathrm{C}(3)-0 \cdot 114, \mathrm{C}(4) 0 \cdot 180$, $\mathrm{O}(2) 0.069$ and $\mathrm{O}(5)-0.076 \AA$. This ring twisting is further evidenced by the torsion angle $\mathrm{C}(31)-\mathrm{C}(3)-$ $\mathrm{C}(4)-\mathrm{C}(41)$ for the cis-substituted $\mathrm{C}(31)$ and $\mathrm{C}(41)$, which has a value of $23.3(2)^{\circ}$. For comparison, torsion angles of $3(1)^{\circ}$ were observed for the cissubstituted H and C atoms at $\mathrm{C}(3)$ and $\mathrm{C}(4)$ in (1b). The exocyclic bond angles at $C(3)$ and $C(4)$ in (1c) are markedly different from one another. Although, as a result of steric contacts, both $\mathrm{C}(31)-\mathrm{C}(3)-\mathrm{C}(4)$ and $C(41)-C(4)-C(3)$ are significantly wider than the equivalent bond angles in ( $1 b$ ), the degree of widening is more pronounced for the latter, which is $119.3(2)^{\circ}$. $\mathrm{C}(41)-\mathrm{C}(4)-\mathrm{C}(5)$ is $112.0(2)^{\circ}, 2.3^{\circ}$ wider than the equivalent angle in (1b), whereas $\mathrm{C}(31)-\mathrm{C}(3)-\mathrm{C}(2)$ is $0.6^{\circ}$ smaller.

## References

Cromer, D. T. \& Liberman, D. (1970). J. Chem. Phys. 53, 1891-1898.
Cromer, D. T. \& Waber, J. T. (1965). Acta Cryst. 18, 104-109.

Gossauer, A. \& Hinze, J.-P. (1981). Hoppe-Seyler's Z. Physiol. Chem. In the press.
Sheldrick, G. M. (1976). SHELX. A progam for crystal structure determination. Univ. of Cambridge. England.
Sheldrick, W. S. (1981). Acta Cryst. B37, 299-300.

## SHORT COMMUNICATION

Contributions intended for publication under this heading should be expressly so marked; they should not exceed about 1000 words; they should be forwarded in the usual way' to the appropriate Co-editor; they will be published as speedily as possible.

Acta Cryst. (1981). B37, 302
Racemic 3-[benzyl(phenyl)phosphinyl]-2-butenoic acid: errata. By M. L. Glówka. Institute of General Chemistry, Technical University of Łódź, 36 Z̈wirki, 90-924 Łódż, Poland
(Received 20 August 1980; accepted 1 September 1980)


#### Abstract

E.s.d.'s for bond lengths and angles, the $y$ coordinate for $C(3)$ [new value $0.7156(9)$ ], and the bond length $\mathrm{C}(16)-\mathrm{C}(11)$ Inew value $1.415(10) \AA$ l in the paper by Glowka [Acta Cryst. (1978), B34, 3089-3091] are revised and a missing hydrogen-bond distance is added $|O(1) \cdots O(2) 2.587(7) \AA|$.


The $y$ coordinate of $\mathrm{C}(3)$ is 0.7156 (9). The remaining corrections are given in Table 1.

The author thanks Professor G. Ferguson for pointing out the errors.

Table 1. Bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$

| $\mathrm{P}-\mathrm{O}(1) \quad 1.4$ | 1.493 (4) | $\mathrm{C}(3)-\mathrm{C}(4) \quad 1$. | 1.415 (14) | $\mathrm{C}(8)-\mathrm{C}(9) \quad 1$ | 1.510 (9) | $\mathrm{C}(13)-\mathrm{C}(14)$ | 1.379 (13) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{P}-\mathrm{C}(1) \quad 1.7$ | 1.796 (6) | $\mathrm{C}(4)-\mathrm{C}(5) \quad 1$. | 1.402 (14) | $\mathrm{C}(9)-\mathrm{O}(2) \quad 1.310$ | 0 (8) | $\mathrm{C}(14)-\mathrm{C}(15) \quad 1.402$ | 02 (13) |
| $\mathrm{P}-\mathrm{C}(7) \quad 1.82$ | 1.827 (7) | C(5)-C(6) 1. | 1.405 (10) | $\mathrm{C}(9)-\mathrm{O}(3) \quad 1.19$ | (8) | $\mathrm{C}(15)-\mathrm{C}(16) \quad 1.40$ | (10) |
| $\mathrm{P}-\mathrm{C}(10) \quad 1.82$ | 1.827 (8) | $\mathrm{C}(6)-\mathrm{C}(1) \quad 1$. | 1.402 (10) | $\mathrm{C}(10)-\mathrm{C}(11) \quad 1.520$ | 0 (8) | $\mathrm{C}(16)-\mathrm{C}(11) \quad 1.4$ | 5 (10) |
| $\mathrm{C}(1)-\mathrm{C}(2) \quad 1.4$ | 1.410 (11) | $\mathrm{C}(7)-\mathrm{C}(8) \quad 1$. | 1.333 (8) | $\mathrm{C}(11)-\mathrm{C}(12) \quad 1.40$ | 4 (10) |  |  |
| $\mathrm{C}(2)-\mathrm{C}(3) \quad 1.4$ | 1.431 (10) | $\mathrm{C}(7)-\mathrm{C}(17) \quad 1.525$ (8) |  | $\mathrm{C}(12)-\mathrm{C}(13) \quad 1.4$ | 4 (10) | $\mathrm{O}(1) \cdots \mathrm{O}(2) \quad 2$. | (7) |
| $\mathrm{P}-\mathrm{C}(7)-\mathrm{C}(8)$ | 113.6 (5) | $\mathrm{O}(1)-\mathrm{P}-\mathrm{C}(7)$ | 113.2 (3) | $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{C}(2)$ | 121.2 (6) | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | 120.0 (7) |
| $\mathrm{P}-\mathrm{C}(7)-\mathrm{C}(17)$ | 117.1 (5) | $\mathrm{O}(1)-\mathrm{P}-\mathrm{C}(10)$ | 112.9 (3) | $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | 127.5 (6) | $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | 121.7 (7) |
| $\mathrm{P}-\mathrm{C}(10)-\mathrm{C}(11)$ | 109.9 (5) | $\mathrm{C}(7)-\mathrm{P}-\mathrm{C}(10)$ | 106.6 (3) | $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(17)$ | 129.3 (6) | $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | 118.9 (8) |
| $\mathrm{P}-\mathrm{C}(1)-\mathrm{C}(2)$ | 116.3 (5) | $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 119.1 (7) | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{O}(2)$ | 117.5 (5) | $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(11)$ | 120.4 (7) |
| $\mathrm{P}-\mathrm{C}(1)-\mathrm{C}(6)$ | 122.5 (5) | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 119.5 (8) | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{O}(3)$ | 119.5 (6) | $\mathrm{C}(16)-\mathrm{C}(11)-\mathrm{C}(12)$ | $120 \cdot 1$ (6) |
| $\mathrm{C}(1)-\mathrm{P}-\mathrm{O}(1)$ | 112.5 (3) | $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 119.9 (7) | $\mathrm{O}(2)-\mathrm{C}(9)-\mathrm{O}(3)$ | 122.9 (7) | $\mathrm{C}(16)-\mathrm{C}(11)-\mathrm{C}(10)$ | 120.7 (6) |
| $\mathrm{C}(1)-\mathbf{P}-\mathrm{C}(7)$ | 105.0 (3) | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $121 \cdot 1$ (7) | $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | 119.2 (6) | $\mathrm{C}(9)-\mathrm{O}(2)-\mathrm{H}(1)$ | 107 (3) |
| $\mathrm{C}(1)-\mathrm{P}-\mathrm{C}(10)$ | $106 \cdot 1$ (3) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(1)$ | 119.2 (7) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | 118.9 (7) | $\mathrm{P}-\mathrm{O}(1) \cdots \mathrm{H}(1)$ | 140 (2) |

0567-7408/81/010302-01\$01.00
(c) 1981 International Union of Crystallography

## International Union of Crystallography

Acta Cryst. (1981). B37, 302

## Submission of papers for publication in Acta Crystallographica

The unhindered selection by authors of the most convenient Co-editor for the submission of papers is one of the policies of Acta Crystallographica stated on the inside front cover of all issues. In recent years, a disproportionate number of
papers have been submitted to Dr P. J. Wheatley and Professor M. M. Woolfson. Many of these papers originated in countries outside the United Kingdom. With the resignations of Professor Woolfson and Dr Wheatley it would be helpful if the authors of such papers, particularly those resident in other European countries, would select a Coeditor not resident in the United Kingdom. Manuscripts receive comparable treatment from all Co-editors, and possible delays in handling or by manuscript transfer will be avoided by a more random manuscript distribution.


[^0]:    * Lists of structure factors, anisotropic thermal parameters and H atom parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35609 ( 14 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography. 5 Abbey Square, Chester CH1 2HU, England.

[^1]:    * Lists of structure factors, anisotropic thermal parameters and H atom parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35610 ( 12 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

